BIOTHERMODYNAMIC DATA CAPTURE

Example: Solubilities of biomaterials in solution

Data source: Carta, R. J. Chem. Thermodyn., 1998, 39, 1038-1048.

General Experiment Description: Solubilities of L-cystine, L-tyrosine, Lleucine, and glycine in sodium chloride solutions at various pH values determined gravimetrically.

Target Properties for the example: Molarities of L-cystine in a complex solution.

Bibliographic information:

No new additions were made to GDC for biothermodynamic data.

See: <u>http://www.trc.nist.gov/GDC.html</u> for general help. See: <u>http://www.trc.nist.gov/helpdocs/basic/BIBLIOGRAPHICinfo.pdf</u> for specific help on entering bibliographic information.

Here is the captured bibliographic information for the example:

🛢 Literature	source description
Help	
Type of docume	ent: Journal article
Title: Sol	ubilities of L-cystine, L-tyrosine, L-leucine, and glycine in sodium chloride solutions at various pH values
Authors: Car	ta, R.[Renzo]
Source: 🔊	. Chem. Thermodyn.@ \$30\$, 379-387
Year:	1998
Key words:	solubility; amino acids; sodium chloride; equilibrium; thermodynamic model
Abstract (if available):	The solubilities of four amino acids L-cystine, L-tyrosine, L-leucine, and glycine.at Ts298 K and at the pH values of 0.00, 1.00, 2.00, 11.00, 12.00, and 13.00 in water solutions of sodium chloride, have been measured. The concentration of NaCl was varied from 0.0 to 3.0. mol.dmy3. The solubility of L-cystine increased as the concentration of salt was raised from 0.0 to 3.0.mol.dmy3. For the same variation of sodium chloride concentration the solubilities of L-tyrosine and L-leucine decreased. The solubility of glycine was not influenced significantly by the presence of NaCl. A simple relation between the ratio of the solubilities of the amino acids with, and without, salt and NaCl concentration is presented. The relation predicts well the solubilities of L-cystine, L-tyrosine, L-leucine, and glycine in aqueous solutions of NaCl starting from known values of the solubilities in pure
	Accept Cancel

After capture of bibliographic info and specification of the reaction components, the Main GDC form looks like this...

В,	Guide	d Data	a Capture - Th	ermophysical	and Thermo	chemical Data				
Eile	<u>E</u> dit	T <u>o</u> ols	Help							
	Refer	ence	Compound	Sample	Mixture	Reaction	Property	Data Tables	Bio	
	1998 ca - L-cy - wat - sod - sod - sod - hyd	ar & 0 ystine Sample ium chli Sample ium hyc Sample rogen c Sample	e 1 (cm,99w%,spl; e 1 (cm;;99.99m%, oride e 1 (cm,99.9w%,e: troxide e 1 (cm;;) shloride e 1 (cm;;)	:) .est) st;;)						
		Г	Next w	ve begii	n to ei	nter the	e Solut	oility in	formatio	n

Initiation of Bio Substance Solubility property capture...



Next...

The **Bio System Properties** form appears for **Solubilities**...

Bio System Help Substance:	m Properties	These fields are pre-filled in accord with the definition of a solubility	
Phase 1 Phase 2 Composition	Solution Crystal Constraints Variables Properties conents present	Description Select or enter text here Sample	▼ Function
	Tabs to navigate between Composition (of the solut Constraints, Variables, a are analogous to those for properties (Example I)	forms for the tion), and Properties r DSC Sample Sample Sample Sample Sample Sample Sample	Function Function
Method of mea	isurement: T d n tł	⁻ his field supports entry of spec lescriptors, such as <i>alpha, beta</i> <i>nonoclinic, amorphous</i> , etc. At p nese are entered as text strings	ial crystal , present, 5.

Define the Composition (of the solution):

Bio System Properties	
Help	
Substance: L-cystine 1. Select	t components present
Phase 1 Solution & Sample	e number (if needed)
Composition Constraints Variables Properties	
Conter components present	
	Next
water	▼ Sample 1 ▼ Function Solvent ▼
sodium chloride	▼ Sample 1 ▼ Function Solvent ▼
sodium hydroxide	Sample Function Solvent
hydrogen chloride	Sample Function Solvent
J	Sample Function
	Sample Function
	Sample Function
]	2 Select the Function of each component
Method of measurement	Solvent
Induced of medicalitation [gravimenic analysis	Buffer component
	Inert
3. Enter the Method of	
Measurement	Next
incusin chieft	

Define the Constraints:

🛱 Bio System Properties		
Help		
Substance: L-cystine		✓ Sample # 1
Phase 1 Solution	-	
Phase 2 Crystal		2. Enter constraint values
Composition Constraints Variables Properties	1. Select constraints	
	↓ _	Next
		Value: 298 K Uncert.
Pressure		Value: 101 kPa Uncert.
	-	Value: Unicert.
-	· •	Value: Uncert.
	- -	Value: Uncert.
	- -	Value: Uncert.
	•	Value: Uncent.
	- -	Value: 7 %
	·	
Method of measurement: gravimetric analysis		Accept Cancel
	2 Enter una	rtainting for constraints
	S. Enter unce	
	if know	n (absolute or percent)

Next tab...

Define the Variables:

🛢 Bio System Properties		
Help		
Substance: L-cystine		✓ Sample # 1 ✓
Phase 1 Solution	_	
Phase 2 Crystal	1 Calast variables	•
Composition Constraints Variables Properties	1. Select variables	
		Next
Var.1 pH	-	Uncert. 🔽 🏹 🎖
Var.2 Solvent: MolaRity	sodium chloride	mol/dm3 Uncert.
	▼	Uncert.
	2 Enter uncertainties	if known
		nt.
		Uncert.
		Uncert.
	_	Uncert.
Method of measurement: gravimetric analysis		Accept Cancel

Next tab...

Define the Properties:

Bio System Properties		
Help		
Substance: L-cystine		✓ Sample # 1 ✓
Phase 1 Solution	Novt to	
Phase 2 Crystal	Description	
Composition Constraints Variables Properties	Data Table	
		<u> </u>
		Next
MolaLity	L-cystine mol/kg	Uncert. 🔽 🏹 🖉 %
1. Select properties (only	1 in the example)	1
Method of measurement: gravimetric analysis	 2. Enter uncertainties for properties Absolute or percent Uncertainties associated with each value can be capture on the next form 	Accept Cancel

Enter numerical values for Variables and Properties:

See page 382 of the example article pdf.

6									
Eile	<u>E</u> dit <u>A</u>	ction <u>H</u> elp							
		Var 1	Var 2	Prop 1 Uno	c1				
		pH	Solvent: Mo	MolaLity (L-c					-
1		0.00	0	0.09412					
	2	0.01	0.5	0.10315				_	
	3	0.01	1.0	0.105.1			ha na sina d		
	- 4	0.02	1.5	0.11864		iumns can	be resized		
	5	0.02	2.0	0.12719		hy "draggin	a" horo		
	6	0.03	2.5	0.14087		by urayyii	ig nere		
	7	0.02	3.0	0.14399				-	
_	8	1.03	0	0.01701					
	9	1.05	0.5	0.02039					
_	10	1.05	1.0	0.02365					
-	11	1.06	1.5	0.02398					
	12	1.05	2.0	0.02401					
	13	1.07	2.5	0.02447					
-	14	1.08	3.0	0.02437					
	10	2.11	05	0.00145					
	10	2.12	1.0	0.00130					
	18	2.14	1.0	0.00172					
	19	2.14	2.0	0.00100					
	20	217	2.0	0.00234					
	21	2.17	3.0	0.00247					
	22	7.81	0	0.00161					
	23	7.81	0.5	0.00169					
	24	7.79	1.0	0.00181	1				-1
1	05	1 7 70		0.00011				194 202	
				Clear the Tal	ble	View plot		Accept	Cancel

See next page...

Ci File E	dit <u>A</u> ction <u>H</u> elp			Colur mak reada	nns are resized to e headings easily able, if necessary.	
		_		1		
	Var.1	Var.2	Prop.1	Unc.1	_	<u> </u>
	pH	Solvent: MolaRity (sodium chloride	e) MolaLity (L-cystine)		_	
1	0.00		0 0.09412	2		
	2 0.01	0.	5 0.10315	1	-	
	3 0.01	1.	0 0.10911		-	
	4 0.02	i	5 U.11864			
	0.02 C 0.02	2.	U U.12713 E 0.14097	,	-	
	7 0.02	2	0 0.14087	1		
	8 1.02		0 0.14333			
	9 1.05	0.	5 0.02039			
	10 1.05	1.	0 0.02365	i		
	11 1.06	1.	5 0.02398			
	12 1.06	2.	0 0.02401			
	13 1.07	2.	5 0.02447	·		
	14 1.08	3.	0 0.02437			
	15 2.11		0 0.00145	i		
	16 2.12	0.	5 0.00156	i		
	17 2.14	1.	0 0.00172	2		
	18 2.14	1.	5 0.00189	1		
Cli	ick View	plot for	0 0.00201		_	
			5 0.00234		Click	Accent
	araphing	options 🖃	0 0.00247			noocpi
L	231 7.81	0	5 0.00161		when	done
	24 7.79	1	0 0.00181			
J	25 7 7 70		E 0.00014		A	
		Clear the Table	View p	plot	Act	cept Cancel

Plotting Options: Plot any property against any variable (2-d only)



Some additional useful features of GDC plotting



Some additional useful features of GDC plotting



